

1,4-*N,C*-Elimination of trialkylsilanol as a new way of fragmentation of *N,N*-bis(trialkylsilyloxy)enamines

A. A. Tishkov, I. M. Lyapkalo, S. L. Ioffe,* Yu. A. Sirelenko, and V. A. Tartakovsky

N. D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences,
47 Leninsky prosp., 117913 Moscow, Russian Federation.
Fax: 007 (095) 135 5328

N,N-Bis(trialkylsilyloxy)enamines (BSENA), obtained recently by the double silylation of nitroalkanes,^{1–3} react with *N*-¹ and *C*-nucleophiles² and rearrange under the action of Lewis acids or upon heating to bistrialkylsilyl derivatives of α -hydroxyoximes.^{1,3}

We found that BSENA containing the $X-CH_2-$ substituent (*X* is the electron-withdrawing group) at the C(2) atom are capable of undergoing a new type of fragmentation under the action of bases: 1,4-*N,C*-elimination of trialkylsilanol to form trialkylsilyl derivatives of unsaturated α,β -oximes.

The new fragmentation of BSENA was performed by the transformation of methyl 4-*N,N*-bis(trimethylsilyloxy)aminobut-3-enonate (3) to methyl 4-trimethylsilyloxyiminobut-2-enonate (4) (Scheme 1). Derivative 4 can be obtained both without isolation of intermediate BSENA 3 under conditions of silylation of the initial nitro compound 1 at a higher temperature (see Scheme 1, stage *c* in comparison with stage *a*), and by treatment of enamine 3 with triethylamine (see Scheme 1, stage *b*).

Desilylation of compound 4 by methanol gives stable methyl 4-hydroxyiminobut-2-enonate (5). Oximes 4 and 5 in $CDCl_3$ exist as a mixture of *E-anti*-, *E-syn*-, and *Z-anti*-isomers (*a*, *b*, and *c*, respectively). The configurations of oximes 5 were established by NMR (Gated, $^1H-^1H$, and $^{13}C-^1H$ correlations) and by the comparison of the chemical shifts and spin coupling constants of

oximes 5 (Table 1) with the corresponding parameters of methylacrylate and isomers of cinnamic acid.⁴

The fragmentation found can be used for the development of a two-stage method for stereoselective β -hydroxyiminoalkylation of compounds with the conjugated double bond.

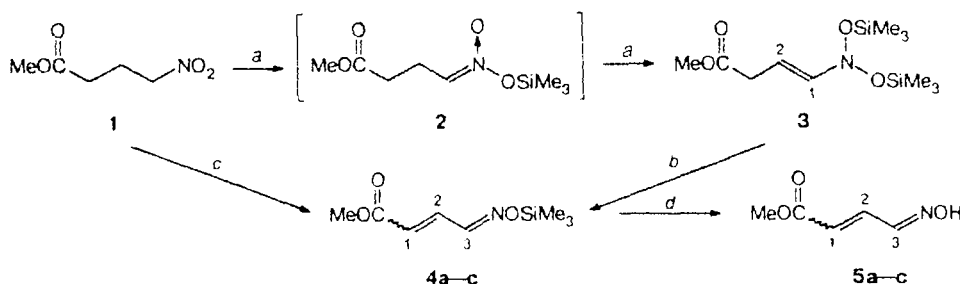
Methyl 4-*N,N*-bis(trimethylsilyloxy)aminobut-3-enonate (3). Yield 67%. 1H NMR ($CDCl_3$, relative to $SiMe_4$), δ : 0.18 (s, 18 H, $SiMe_3$); 3.03 (dd, 2 H, CH_2 , $^3J = 8$ Hz, $^4J = 1$ Hz); 3.70 (s, 3 H, OMe); 5.60 (dt, 1 H, CH , $^3J = 14$ Hz); 6.06 (dd, 1 H, $CH=N$). ^{13}C NMR ($CDCl_3$, relative to $SiMe_4$), δ : 0.88 ($SiMe_3$); 35.21 (CH_2); 52.79 (OMe); 114.29 (CH); 145.95 ($CH=N$); 172.42 ($C=O$). ^{29}Si NMR (INEPT, $CDCl_3$, $SiMe_4$), δ : 24.39.

Methyl 4-trimethylsilyloxyiminobut-2-enonate (4). Yield 71% (calculated per nitro compound 1 at stage *c*), b.p. 46–50 °C (0.5 Torr), n_D^{20} 1.4760. The ratio **4a** : **4b** : **4c** = 4.0 : 1.0 : 1.3 (according to 1H NMR spectra).

Methyl 4-hydroxyiminobut-2-enonate (5). Yield 97%, m.p. 92–94 °C (from CH_2Cl_2 –pentane). The ratio **5a** : **5b** : **5c** = 4.0 : 1.0 : 1.3 (according to 1H NMR spectra). Found (%): C, 45.79; H, 5.26; N, 10.97. $C_5H_7NO_3$. Calculated (%): C, 46.51; H, 5.47; N, 10.85.

This work was performed at the Scientific Educational Center of the Institute of Organic Chemistry of the Russian Academy of Sciences and in the Moscow Chemical Lyceum with the financial support of the Russian Foundation for Basic Research (Project No. 96-03-32472).

Scheme 1



Reagents and conditions: *a*, Me_3SiBr/Et_3N (1 : Me_3SiBr : Et_3N = 1 : 2.03 : 2.02, mol/mol) in $ClCH_2CH_2Cl$, –28 °C, 96 h; *b*, Et_3N in $ClCH_2CH_2Cl$, –28 °C; *c*, Me_3SiBr/Et_3N (1 : Me_3SiBr : Et_3N = 1 : 3.3 : 3.4, mol/mol) in $ClCH_2CH_2Cl$, –15 °C → 10 °C, 48 h; *d*, MeOH (excess), NH_4F , 15 °C.

Table 1. Parameters of ^1H , ^{13}C , and ^{29}Si NMR spectra (INEPT) for products 4 and 5 (in CDCl_3)

Pro- duct	Con- forma- tion	$\delta\ ^1\text{H}^a$ ($^3J_{\text{H,H}}/\text{Hz}$)					$\delta\ ^{13}\text{C}^a$ ($^3J_{\text{H,H}}/\text{Hz}$)					$\delta\ ^{29}\text{Si}^a$	
		H(1) (d)	H(2) (dd)	H(3) (d)	MeO (s)	SiMe ₃ (s)	C(1)	C(2)	C(3)	MeO	C=O		SiMe ₃
4a	<i>E-anti</i>	6.15	7.37, 10	7.93, 16	3.78	0.27	126.37	138.04	152.87	51.63	166.06	-1.08	28.25
4b	<i>E-syn</i>	6.17	7.86, 10	7.42, 17	3.80	0.27	127.22	129.96	149.60	51.77	166.10	-1.08	27.78
4c	<i>Z-anti</i> ($^4J = 1$) ^b	6.01	6.69, 10	8.95, 12	3.76	0.27	122.56	137.49	152.36	51.36	166.01	-1.08	27.61
5a	<i>E-anti</i>	6.16	7.34, 10	7.87, 16	3.75	—	126.04 (dt), ^c 164	138.35 (dd), ^c 161	149.05 (dd), ^c 166	51.97	166.78	—	—
5b	<i>E-syn</i>	6.18	7.84, 10	7.26, 16	3.76	—	127.21 (dt), ^c 164	129.92 (dd), ^c 165	145.86 (dd), ^c 178	51.99	166.78	—	—
5c	<i>Z-anti</i> ($^4J = 1$) ^b	6.01	6.69, 9	8.88, 13	3.74	—	122.44 (d), ^c 166	137.59 (dd), ^c 159	148.52 (dd), ^c 173	51.66	166.78	—	—

^a Relative to SiMe_4 .^b The existence of the constant 4J is additional evidence for the Z-configuration of the C=C bond.⁵^c The multiplicity and spin coupling constants of ^{13}C NMR signals were determined by the Gated procedure.

References

1. H. Feger and G. Simchen, *Lieb. Ann. Chem.*, 1986, 1456.
2. I. M. Lyapkalo, S. L. Ioffe, Yu. A. Strelenko, and V. A. Tartakovsky, *Izv. Akad. Nauk, Ser. Khim.*, 1995, 1182, 1183 [*Russ. Chem. Bull.*, 1995, **44**, 1142 (Engl. Transl.)].
3. H. Feger and G. Simchen, *Lieb. Ann. Chem.*, 1986, 428.
4. Hesse, H. Meier, and B. Zech, in *Spektroskopie Methoden in der organischen Chemie*, Georg Thieme Verlag, Berlin, 1995, 122, 158, 200, 214.
5. A. Gordon and R. Ford, in *The Chemist's Companion. A Handbook of Practical Data, Techniques, and References*, John Wiley and Sons, New York—London, 1972.

Received November 28, 1996;
in revised form December 14, 1996